

PHYSICS

41. (1)

42. (4)

43. (3)

44. (1)

45. (2)

Displacement current

$$I_d = C \frac{dV}{dt} = C \frac{d}{dt}(V_0 \sin \omega t)$$

46. (3)

Using, $\lambda = \frac{12.27}{\sqrt{V}} \text{ Å}$

$$\Rightarrow \sqrt{V} = \frac{12.27 \times 10^{-10}}{1.227 \times 10^{-11}} = 10^2$$

[Given $\lambda = 1.227 \times 10^{-11} \text{ m}$]

$$\therefore V = 10^4 \text{ volt}$$

47. (4)

$$\text{Power, } P = \frac{E}{t} = \frac{n}{t} \left(\frac{hc}{\lambda} \right)$$

$$\therefore \text{Number of protons/second, } \frac{n}{t} = \frac{P}{\left(\frac{hc}{\lambda} \right)}$$

$$\frac{n}{t} = \frac{3.3 \times 10^{-3} \times 6 \times 10^{-7}}{6.6 \times 10^{-34} \times 3 \times 10^8} = 10^{16}$$

48. (4)

From Einstein's photoelectric equation

$$\frac{hc}{\lambda} = \phi_0 + k$$

where ϕ_0 = work function

k = maximum kinetic energy of photoelectrons

$$\therefore \frac{hc}{\lambda} = k = \frac{P^2}{2m}$$

$$\Rightarrow P = \sqrt{\frac{2mhc}{\lambda}} [\because \phi = \text{negligible}]$$

de-Broglie wavelength,

$$\lambda_d = \frac{h}{p} = \frac{h}{\sqrt{\frac{2mhc}{\lambda}}} \Rightarrow \sqrt{\lambda} = \lambda_d \sqrt{\frac{2mc}{h}}$$

$$\therefore \lambda = \left(\frac{2mc}{h} \right) \lambda_d^2$$

49. (2)

$$(E_2 - E_1) = hv = \frac{hc}{\lambda}$$

$$\therefore \frac{hc}{\lambda_1} = (E_C - E_B), \frac{hc}{\lambda_2} = (E_B - E_A) \text{ and}$$

$$\frac{hc}{\lambda_3} = (E_C - E_A)$$

$$\text{Now, } (E_C - E_A) = (E_C - E_B) + (E_B - E_A)$$

$$\text{or } \frac{hc}{\lambda_3} = \frac{hc}{\lambda_1} + \frac{hc}{\lambda_2} \text{ or } \frac{1}{\lambda_3} = \frac{1}{\lambda_1} + \frac{1}{\lambda_2}$$

$$\therefore \frac{1}{\lambda_3} = \frac{\lambda_1 + \lambda_2}{\lambda_1 \lambda_2} \text{ or } \lambda_3 = \frac{\lambda_1 \lambda_2}{\lambda_1 + \lambda_2}$$

50. (1)

Gain in BE = (BE) of products - (BE) of reactants.

$$= [120 + 120] \times 8.5 - [240] \times 7.6$$

$$= (240) \times 8.5 - 240 \times 7.6$$

$$= (2040 - 1824) \text{ MeV} = 216 \text{ MeV}$$